

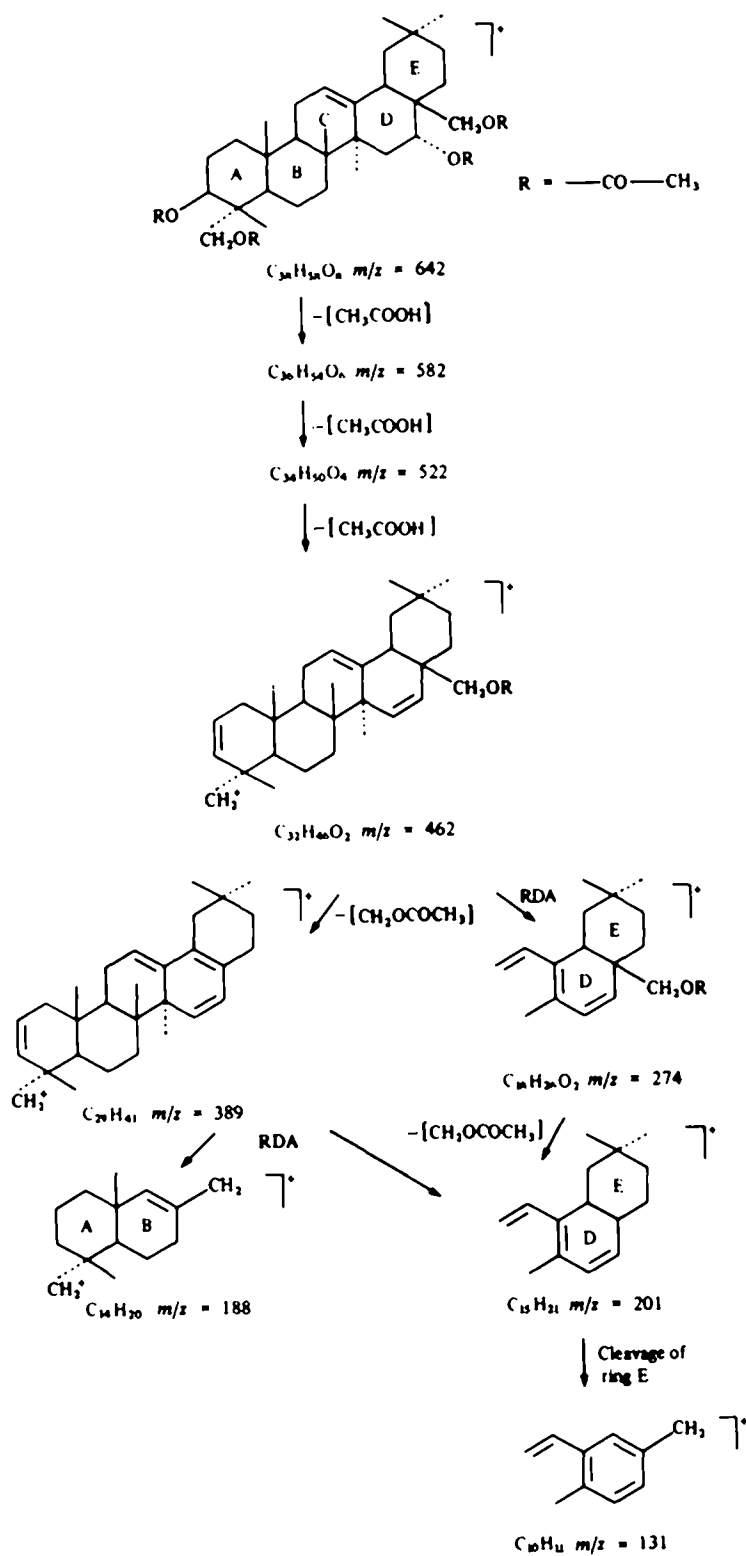


Table 1.  $^{13}\text{C}$ NMR chemical shifts of aglycone 3, its acetate 3a, saponin 2 and glycosidation shifts  $\Delta\delta$  (in parentheses in ppm)

	3	3a	2	$\Delta\delta$
C-1	38.9	38.1	39.2	(+0.3)
C-2	27.4	23.3	25.9	(-1.5)
C-3	74.1	74.5	82.5	(+8.4)
C-4	42.7	41.4	43.6	(+0.9)
C-5	48.9	48.4	47.9	(-1)
C-6	18.6	18.1	17.7	(-0.9)
C-7	33.0	32.7	32.9	(-0.1)
C-8	40.1	40.1	42.4	(+2.3)
C-9	47.2	47.1	50.6	(+3.4)
C-10	37.1	36.9	36.8	(+0.3)
C-11	23.8	23.6	19.3	(-4.5)
C-12	122.3	122.3	36.9	(-85.4)
C-13	145.2	142.3	86.4	(-58.8)
C-14	42.0	41.0	44.6	(+2.6)
C-15	34.7	31.3	34.2	(-0.5)
C-16	73.9	75.8	77.5	(+3.6)
C-17	40.9	38.3	44.6	(+3.7)
C-18	42.6	42.4	51.5	(+8.9)
C-19	48.2	47.4	38.9	(-9.3)
C-20	31.1	30.7	31.8	(+0.7)
C-21	37.1	35.9	37	(-0.1)
C-22	30.2	30.0	31.8	(+1.6)
C-23	68.6	65.6	67.4	(-1.2)
C-24	12.9	13.2	13.2	(+0.3)
C-25	16.2	16.1	17	(+0.8)
C-26	17.1	16.9	19.5	(+2.4)
C-27	27.3	26.7	18.6	(-8.7)
C-28	70.2	71.4	78	(+7.8)
C-29	33.3	33.3	33.7	(+0.4)
C-30	24.9	24.3	24.7	(-0.2)
O-CO-Me		170.8		
		170.6		
		170.5		
		170.0		
O-CO-Me		21.7		
		21.0		
		20.6		
C-1'	Xyl		107.4	
C-2'			80.1	
C-3'			78.0	
C-4'			71.4	
C-5'			64.9	
C-1''	Glc		103.9	
C-2''			77.2	
C-3''			85.2	
C-4''			78.3	
C-5''			78.2	
C-6''			62.7	
C-1'''	Glc		103.6	
C-2'''			76.0	
C-3'''			78.2	
C-4'''			71.0	
C-5'''			77.7	
C-6'''			61.3	
C-1''''	Ara		104.9	
C-2''''			73.3	
C-3''''			75.95	
C-4''''			70.7	
C-5''''			64.2	

primary acetoxy groups, an AB system at  $\delta$ 3.94 and 4.26 ( $J = 11$  Hz, 23  $\text{CH}_2\text{OAc}$ ) [5] and a singlet at 4.01 (28  $\text{CH}_2\text{OAc}$ ). The signal at 5.03 ( $t$ -like) in the  $^1\text{H}$  NMR spectrum of 3a could be attributed to the 3 $\alpha$ -proton geminal to the 3 $\beta$ -hydroxyl group [12-14]. The mass spectrum of 3 and 3a confirmed the  $\Delta^{12}$ -oleanene structure. The typical retro-Diels-Alder (RDA) fragmentation of ring C [15] resulted in characteristic fragment ions as shown in Scheme 1. Therefore, compound 3 was concluded to be 3 $\beta$ ,16 $\alpha$ ,23,28-tetrahydroxyolean-12-ene. This compound is closely related to saponogenins extracted from other Primulaceae [14, 16] and similar to 23-hydroxyprimulagenin A isolated from *Bupleurum falcatum* (Umbelliferae) [6].

The negative ionization FAB mass spectrum of the saponin 2 showed the molecular peak  $[\text{M} - \text{H}]^-$  at  $m/z$  1061. Glucose, xylose and arabinose were found in an acid hydrolysate and partial hydrolysis on TLC showed glucose and arabinose as terminal sugars. Since the molecular weight of the saponogenin was 474, the relative amounts of the sugars were xylose: 1, arabinose: 1, glucose: 2. This was in agreement with the mass spectral fragmentation pattern with peaks at  $m/z$  (rel. int.) 1061  $[\text{M} - \text{H}]^-$  (100); 929  $[\text{M} - \text{Ara}]^-$  (16); 899  $[\text{M} - \text{Glc}]^-$  (10); 767  $[\text{M} - \text{Ara} - \text{Glc}]^-$  (21); 605  $[\text{M} - \text{Ara} - 2 \text{ Glc}]^-$  (15); 473  $[\text{M} - \text{Ara} - 2 \text{ Glc} - \text{Xyl}]^-$  (20). The simultaneous losses of fragment 133 (arabinose) and fragment 162 (glucose) were indicative of two terminal sugars: an arabinose moiety and a glucose moiety, suggesting two sites of attachment or a branched side-chain. The positive ionization FAB MS confirmed this result with peaks at  $m/z$  (rel. int.) 1147  $[\text{M} + 2\text{Na} + \text{K}]^+$  (1.6), 1108  $[\text{M} + 2\text{Na}]^+$  (15), 1085  $[\text{M} + \text{Na}]^+$  (100), 953  $[(\text{M} + \text{Na}) - \text{Ara}]^+$  (11), 923  $[(\text{M} + \text{Na}) - \text{Glu}]^+$  (9), 791  $[(\text{M} + \text{Na}) - \text{Glu} - \text{Ara}]^+$  (11), 628  $[(\text{M} + \text{Na}) - \text{Ara} - 2 \text{ Glu}]^+$  (16), 473  $[(\text{M}) - \text{Xyl} - 2 \text{ Glu} - \text{Ara}]^+$  (5).  $^{13}\text{C}$ NMR chemical shifts (250 MHz,  $\text{C}_3\text{D}_3\text{N}$ , TMS as internal standard) are shown in Table 1. The 'up-down'  $^{13}\text{C}$ NMR spectrum did not give signals at  $\delta$ 122 and  $\delta$ 144 but showed seven quaternary carbons (six for compound 3). The most important changes affected C-12, C-13, C-19 and C-27 which were shifted upfield whereas the C-14, C-16, C-17, C-18 and C-28 signals were shifted downfield. These chemical shifts were very close to those indicated for compounds which possess an oxide ring involving C-28 and C-13, for example androsacenol [17], saikogenin G [4], saikosaponin d [6] and saikosaponin from *Bupleurum longeradiatum* [18]. Moreover, the IR spectrum of 2 showed a sharp band at  $890 \text{ cm}^{-1}$  characteristic of an ether linkage [13, 17]. This suggested a 13 $\beta$ ,28-oxide structure for the aglycone moiety of 2. On acid hydrolysis such a bridged system which has been described in Primulaceae [13, 17] and Umbelliferae [9, 18] would be readily converted to a  $\Delta^{12}$ -17- $\text{CH}_2\text{OH}$  moiety [17] which would explain the structure of 3. Thus, the structure of the aglycone moiety of saponin 2 was established as 3 $\beta$ ,16 $\alpha$ ,23-trihydroxy-13 $\beta$ ,28-epoxyoleanane or 23-hydroxyprotoprimulagenin A. The shifts of C-3, C-16 and C-23 of saponin 2 were compared with signals available for the corresponding positions in compound 3. The single C-3 signal was found to be shifted downfield by 8.4 ppm, indicating the site of glycosidation. Consequently, saponin 2 would possess only one saccharidic chain which is branched and located at C-3. The assignment of carbon signals due to the common sugar moiety of 2 was carried out by comparison with those



reported for the saccharide chain of related saponins [7, 19–22], chemical shifts of pure sugars and previous knowledge of glycosidation effects. The anomeric carbon signals at  $\delta$ 107.4, 103.9, 103.6 and 104.9 confirmed the presence of four monosaccharide units which are substituted on C-1. Thus, the carbon chemical shifts of the terminal arabinose and of the terminal glucose could be attributed. The inner glucose unit must not be linked at C-6 since the signal at  $\delta$ 62 which would correspond to the  $-\text{CH}_2\text{OH}$  group of an unsubstituted glucopyranoside was present. C-3' and C-4' of the inner glucose were shifted downfield by ca 7 ppm indicating glycosidic linkages but we did not discriminate which one of the two terminal sugars was 1  $\rightarrow$  3 and 1  $\rightarrow$  4 linked. In the same way, the signal at  $\delta$ 80.1 was attributed to C-2' of the inner xylose bonded to the anomeric carbon of the inner glucose. For the four sugars, the distribution of chemical shifts was in agreement with a pyranoside structure. Therefore, the saponin isolated from *Anagallis arvensis* would be the 3-O-glucose (1  $\rightarrow$  3 or 1  $\rightarrow$  4)-[arabinose (1  $\rightarrow$  4 or 1  $\rightarrow$  3)]-glucose (1  $\rightarrow$  2)-xyloside of 23-hydroxyprotoprimulagenin A.

Incubation of compound 1 with  $\beta$ -glucosidase afforded glucose together with saponin 2. Negative FAB-MS of saponin 1 gave a well-defined molecular ion at  $m/z$  1223. The fragmentation pattern shown in Table 2 involved successive expulsions of sugar molecules. The first signals at  $m/z$  1091 (12%) and 1062 (12%) corresponded to the simultaneous losses of a terminal arabinose and a terminal glucose. The other important ions were at  $m/z$  767 (20%) and 605 (80%) showing the further elimination of two glucose and one xylose units to give aglycone peak  $[\text{M} - \text{H}]^-$  at  $m/z$  473. By comparison with 2, this fragmentation mechanism permitted the assignment of the site of the additional glucose on the glucosyl moiety but not on the arabinosyl moiety. Positive FAB-MS was in agreement with this result, the molecular peak  $[\text{M} + \text{Na}]^+$  appeared at  $m/z$  1247. In a comparison of the  $^{13}\text{C}$  NMR spectra of compounds 1 and 2, signals due to the aglycone moiety of 1 were superimposable on those of 2, which confirmed the identity of the aglycone moiety of both saponins. Thus, 1 could be formulated as the 3-O-glucose-glucose-[arabinose]-glucose-xyloside of 23-hydroxyprotoprimulagenin A.

#### EXPERIMENTAL

NMR spectra were taken in  $\text{C}_5\text{D}_5\text{N}$  using TMS as an internal standard:  $^1\text{H}$  NMR at 60 MHz,  $^{13}\text{C}$  NMR at 20 MHz for sapogenin and its acetyl derivative (3 and 3a),  $^{13}\text{C}$  NMR at 250 MHz for saponins 1 and 2. Mass spectra of 3 and 3a were recorded by direct inlet at 70 eV ionization. Mass spectra of saponins 1 and 2 were measured by FAB-MS using a ZAB HF

combined with a data system. The solvent used was PEG 400. Spectra in positive ionization and in negative ionization were recorded.

**Plant material.** *Anagallis arvensis* was collected in the suburbs of Rennes, France, in Summer.

**Extraction and isolation of saponins.** Dried and milled whole plant was exhaustively defatted and depigmented with  $\text{CHCl}_3$ , then repeatedly extracted with MeOH at room temp. The MeOH extract was evaporated to dryness and a soln of the residue (yield 19.4%) in  $\text{H}_2\text{O}$  was washed with EtOAc and then extracted  $\times$  3 with 1-BuOH saturated with  $\text{H}_2\text{O}$ . The BuOH was removed *in vacuo* affording crude saponin extract (yield 4.1%) which was subjected to CC on silica gel (70–230 mesh Merck) with  $\text{CHCl}_3$  MeOH (1:1) as the eluant. The major saponins 1 and 2 were further submitted to droplet counter current chromatography (DCCC, Rikakikay Co Ltd, DCCC-A);  $\text{CHCl}_3$  MeOH-*n*-PrOH- $\text{H}_2\text{O}$  (9:12:1:8) ascending mode. Eluates chromatographed on silica gel TLC ( $\text{CHCl}_3$ -MeOH  $\text{H}_2\text{O}$  65:25:4) gave analytical material.

**Acid hydrolysis of the saponins.** Saponin 1 or 2 (1 g) in 0.5 N HCl (100 ml) was heated on a steam bath for 90 min, cooled, diluted with  $\text{H}_2\text{O}$  and extracted  $\times$  3 with  $\text{CHCl}_3$  MeOH (9:1). The  $\text{H}_2\text{O}$  layer was lyophilized and monosaccharides analysed by TLC on cellulose, solvents: EtOAc  $\text{C}_5\text{H}_5\text{N}$ - $\text{CH}_3\text{COOH}$ - $\text{H}_2\text{O}$  (36:36:7:21) and  $\text{H}_2\text{O}$  PhOH (1:5). Glucose, xylose and arabinose were identified. The  $\text{CHCl}_3$  extracts were combined, washed with  $\text{H}_2\text{O}$  and dried. Isolation of the sapogenin (compound 3) was carried out using a silica gel column and eluting with  $\text{CHCl}_3$  MeOH (9:1).

**Compound 3.** MS  $m/z$  (rel. int.): 474  $[\text{M}]^+$  (12), 456 (8), 250 (32), 232 (20), 223 (28), 219 (100), 201 (60), 189 (20), 131 (24); IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3400, 2940, 1640, 1450, 1380, 1045, 1000, 760;  $^1\text{H}$  NMR ( $\text{C}_5\text{D}_5\text{N}$ ):  $\delta$ 1.03, 1.08, 1.12 (15 H, s, 5  $\times$  Me); 1.77 (3 H, s, Me) 2.59 (1 H, br s, 18-H), 3.69 (1H, s, 3 $\alpha$ -H), 3.69, 3.74, 4.08, 4.22 (4H, m, 23-H), 4.59 (2H, br s, 16 $\beta$ -H), 5.4 (1H, s, 12-H).  $^{13}\text{C}$  NMR data: see Table 1.

**Acetylation of 3.** The aglycone 3 (300 mg) was refluxed for 2 hr with  $\text{Ac}_2\text{O}$  (10 ml) in pyridine (10 ml). The reaction mixture was evaporated to dryness. Column chromatography of the tetraacetate on silica gel eluting with  $\text{CHCl}_3$  MeOH (95:5) gave pure compound 3a. MS  $m/z$  (rel. int.): 642  $[\text{M}]^+$  (2), 582 (6), 522 (20), 462 (4), 389 (8), 274 (22), 201 (100), 188 (18), 131 (20); IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 2940, 1760, 1450, 1380, 1240, 1045, 780;  $^1\text{H}$  NMR:  $\delta$ 0.84, 0.92, 0.98 (5  $\times$  Me), 1.43 (Me), 2.03, 2.13 (4  $\times$   $\text{H}$ , s,  $\text{CH}_3\text{COO}$ ) 3.94, 4.26 (AB system,  $J = 11$  Hz, 23  $\text{CH}_2\text{OAc}$ ) 4.01 (2H, s, 28- $\text{CH}_2\text{OAc}$ ), 5.03 (1H, *r*-like, 3 $\alpha$ -H) 5.35 (1H, br s, 12-H) 5.49 (1H, br s, 16 $\beta$ -H).  $^{13}\text{C}$  NMR data: see Table 1.

**Partial acid hydrolysis on TLC [23].** Saponins 1 or 2 were applied on silica gel TLC and left in an HCl atmosphere at room temp. for 60 min. HCl vapour was eliminated under hot ventilation then authentic sugars were applied to the plate. The plate was developed with the solvent system  $\text{CHCl}_3$  MeOH  $\text{H}_2\text{O}$  (6.4:4:0.8) and spots detected by spraying:

Table 2. Negative ion FAB-mass spectrum of saponin 1

$m/z$	Rel. int.	Interpretation
1223	100	$[\text{M} - \text{H}]^-$ (for $\text{C}_{58}\text{H}_{96}\text{O}_{27}$ )
1091 (1223 – 132)	12	$[\text{M} - \text{Ara}]^-$
1062 (1223 – 162)	12	$[\text{M} - \text{Glc}]^-$
767 (1223 – 132 – 2 $\times$ 162)	20	$[\text{M} - \text{Ara} - 2 \text{ Glc}]^-$
605 (1223 – 132 – 3 $\times$ 162)	80	$[\text{M} - \text{Ara} - 3 \text{ Glc}]^-$
473 (1223 – 2 $\times$ 132 – 3 $\times$ 162)		$[\text{M} - \text{Ara} - 3 \text{ Glc} - \text{Xyl}]^- = \text{aglycone}$

aniline-diphenylamine- $\text{H}_3\text{PO}_4$ -MeOH (2:2:10:96), followed by heating. For both saponins two terminal sugars appeared which were identified as glucose and arabinose.

**Enzymatic hydrolysis of 1.** A mixture of 1 (3 mg),  $\beta$ -glucosidase (3 mg) and phosphate buffer pH 5 (0.75 ml) was left to stand at 37° for 24 hr. Control TLC using BuOH-HOAc- $\text{H}_2\text{O}$  (6:1:2) as eluant showed that compound 1 gave saponin 2 together with glucose.

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